

1. Convert the following to exponents of 10:

a.  $4.47 \times 10^9 = \underline{10^{9.65}}$

b.  $6.022 \times 10^{23} = \underline{\hspace{2cm}}$

c. 0.21 atm =  $\underline{\hspace{2cm}}$

2. Convert the following to scientific notation:

a.  $10^{4.4} = \underline{\hspace{2cm}}$

b.  $10^{3.14159} = \underline{\hspace{2cm}}$

c.  $10^{-10.4} = \underline{\hspace{2cm}}$

d.  $10^{-1.6} = \underline{\hspace{2cm}}$

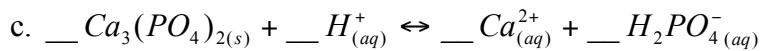
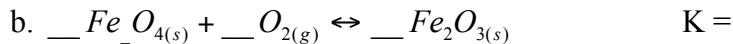
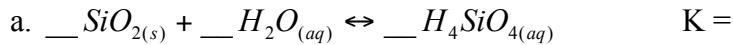
3. Provide the results of these operations in both exponents of 10 and scientific notation:

a.  $(10^{-8.4} \times 10^{-6.4})^{\frac{1}{2}} = \underline{\hspace{2cm}} = \underline{\hspace{2cm}}$

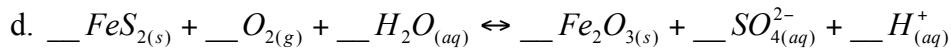
b.  $\frac{10^{-1.2} \times 10^{-11.6}}{10^{-2.4}} = \underline{\hspace{2cm}} = \underline{\hspace{2cm}}$

c.  $\frac{(4.67 \times 10^{-4})(1.6 \times 10^{-9})}{3.85 \times 10^{-12}} = \underline{\hspace{2cm}} = \underline{\hspace{2cm}}$

4. Balance the equations and write expressions for equilibrium constants:



K =



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