Transport Modeling of Wastewater in the Sana'a Aquifer

Bryce DeGuise Geology 428/628 December 9, 2014

Outline

- Introduction
- Background Information
- Concern
- Original Study
- Transport study
- Results
- Future Research Possibilities
- Questions
- References

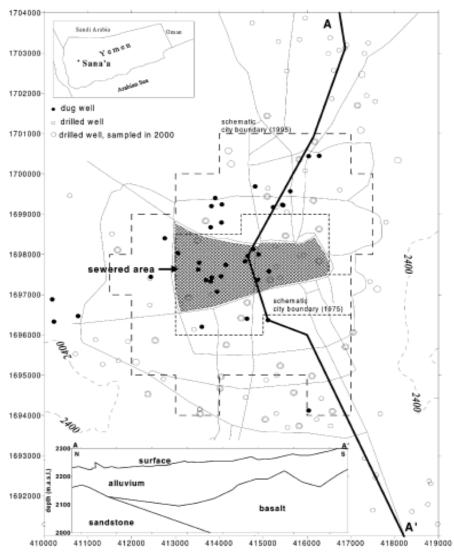
Introduction

• Sana'a Aquifer in Yemen



Background Information

- Sana'a Basin
 - Area = 3,200 km^2
 - 2,200 m above sea level
 - Geological Formations:
 - Quaternary Alluvium
 - Max thickness = 200 m
 - Tertiary Basalt
 - Thickness = 800 m (south)
 - Cretaceous Sandstone
 - Average thickness = 300 m



Background Information Continued

- Focus is on the unconsolidated alluvium
- Shallow water table approximate depth of 40 m (which is heavily exploited)

Alluvium

- Consists of mainly basaltic fragments
- Also contains fractions of clay, silt, sand and pebble sized grains
- Wastewater infiltrates into this part of the aquifer

Concern

- Population growth
 - Average 6.1 percent annual growth from (1997-2001)
 - 1994 total population around 1,000,000
 (2004 estimated population of 1.708 million)

Wastewater

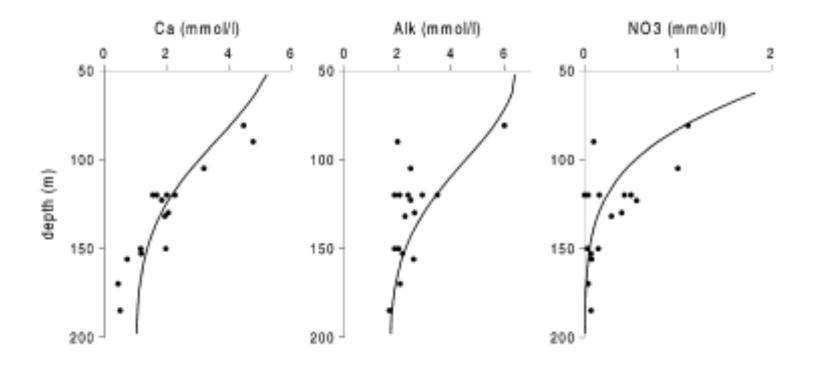
- Disposal differs from the that of U.S.
 - Sana'a uses sewage ponds, cesspits and old water wells for disposal of solid and liquid wastewater

Original Study

- Element quantities at different depths
- Examined different wells
 - Dug wells (before 1970's) and bored wells (1960's and after)
- Valuation of oxidation, nitrification difference with time (samples were taken in 1995 and then in 2000)

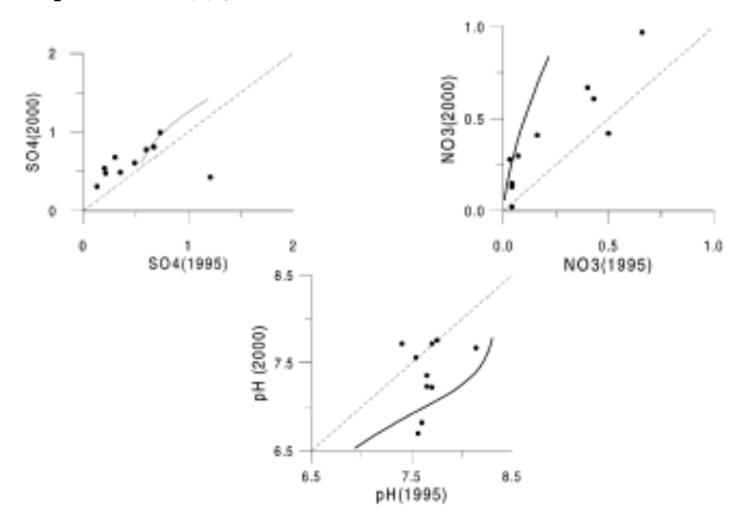
Original Study Continued

- Major cations and anions compared to depth
- Solid lines are model results from PHREEQC



Original Study Continued

• Comparison of 1995 to 2000 collected data



Transport Study

- Based off information given affects through the saturation zone given initial report was not shown.
- The purpose is see how the elements would react as they pass through the soil.
- Multiple wells showed contamination in initial study.

Transport Study

• <u>Hypothesis</u>:

- Chlorine concentrations will go up as the system proceeds through the each step.
- Calcium and potassium would increase concentrations.

PHREEQC Information

Wastewater Input

Alluvium Water Input

```
SOLUTION O Wastewater - Raw sewage
                                        SOLUTION 1-40 Initial Alluvium water
         23
temp
                                                   26
                                           temp
рΗ
         7.3
                                           рΗ
                                              8.35
pe
                                              12
                                           pe
redox
         pe
                                           redox
                                                   pe
units
         mmol/1
                                           units
                                                   mmol/1
density 1
                                           density 1
Alkalinity 15.7
                                           Alkalinity 1.65
Ca
         2.5
                                           Ca
                                                   2.5
Cl
         10
                                              1.52
      1.5
                                               0.07
Mg
         1.5
                                              0.3
                                           Μg
N(-3)
         10
                                           N(-3)
                                                   0
Na
      9.82
                                           Na
                                                  1.6
S(6) 1.81
                                           S(6) 1.81
        1 # kg
water
                                                   1 # kg
                                           water
```

Note: Information was given from research paper

Saturation Indices (SI)

Wastewater

Phase	SI**	log IAP	log K(296 K,	1 atm)
Anhydrite	-1.83	-6.08	-4.26	CaSO4	
Aragonite	0.44	-7.89	-8.32	CaCO3	
Calcite	0.58	-7.89	-8.47	CaCO3	
CO2 (g)	-1.39	-2.83	-1.44	CO2	
Dolomite	1.05	-15.99	-17.04	CaMg (CC	3)2
Gypsum	-1.50	-6.08	-4.58	CaSO4:2	2H2O
H2 (g)	-18.65	-21.74	-3.10	H2	
H2O(g)	-1.55	-0.00	1.55	H20	
Halite	-5.73	-4.16	1.57	NaCl	
NH3 (g)	-5.93	-4.09	1.84	NH3	
02 (g)	-46.67	-49.55	-2.88	02	
Sylvite	-5.87	-4.98	0.89	KC1	

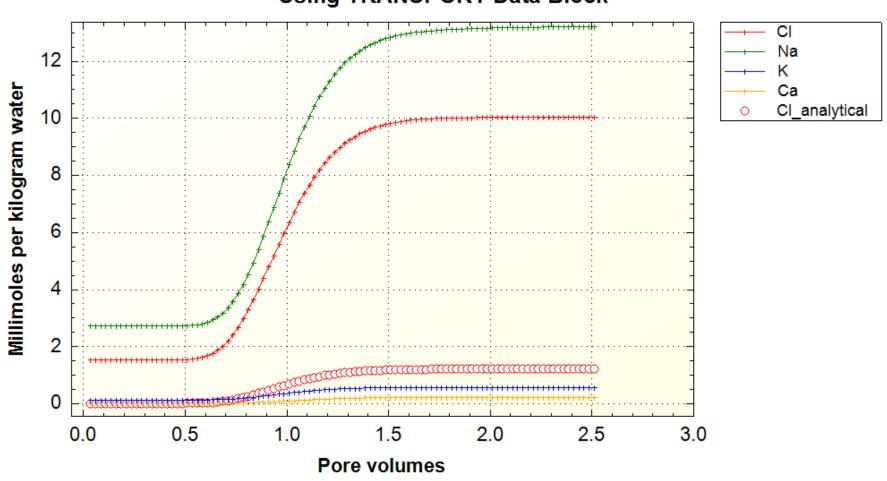
Saturation Indices (SI)

Alluvium Water

Phase	SI**	log IAP	log K(299 K,	1 atm)
Anhydrite	-1.55	-5.84	-4.29	CaSO4	
Aragonite	0.67	-7.68	-8.34	CaCO3	
Calcite	0.81	-7.68	-8.49	CaCO3	
CO2 (g)	-3.40	-4.87	-1.47	CO2	
Dolomite	0.83	-16.29	-17.11	CaMg (CO3	3) 2
Gypsum	-1.26	-5.84	-4.58	CaSO4:2H	120
H2 (g)	-40.75	-43.85	-3.10	H2	
H2O(g)	-1.48	-0.00	1.48	H20	
Halite	-7.28	-5.71	1.57	NaCl	
02 (g)	-1.45	-4.35	-2.90	02	
Sylvite	-7.97	-7.07	0.90	KC1	

Transport Model

Using TRANSPORT Data Block



Results

- Based of the transport graph, that all element concentrations increased.
- I noticed that the SI of wastewater and the alluvium, had a higher concentrations of dolomite than calcite.
- In the tests conducted within the paper indicate the calcite was dominate SI to precipitate out of solution at the levels.

Future Test Possibilities

• Obtain more samples from the future date and compare to previous tests to see if conditions have changed. This includes any industrial, or additional sewer treatment systems since first test.

Questions?

Reference

• Foppen, J.W.A. (2002), Impact of high-strength wastewater infiltration on groundwater quality and drinking water supply: the case of Sana'a Yemen. Journal of Hydrology. 263, 198-216.