

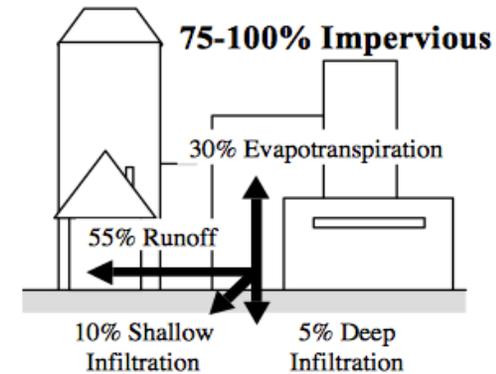
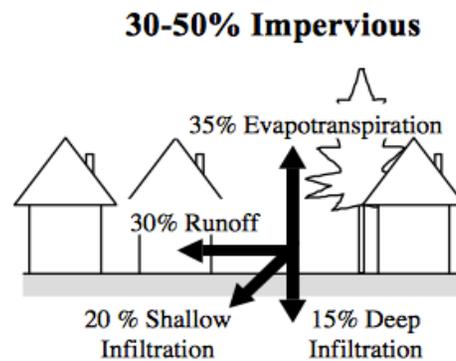
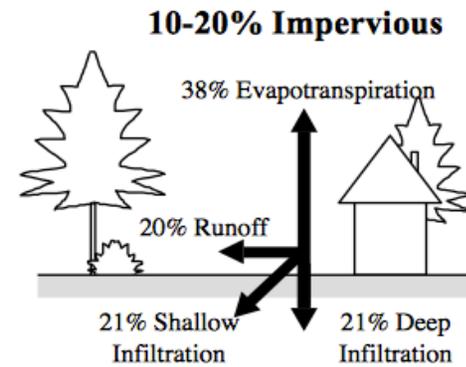
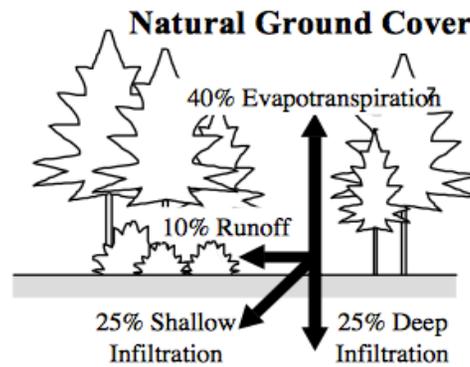
Lead Contamination in Urban Surface Waters

{ Tyler Kuehn

NDSU Geochemistry

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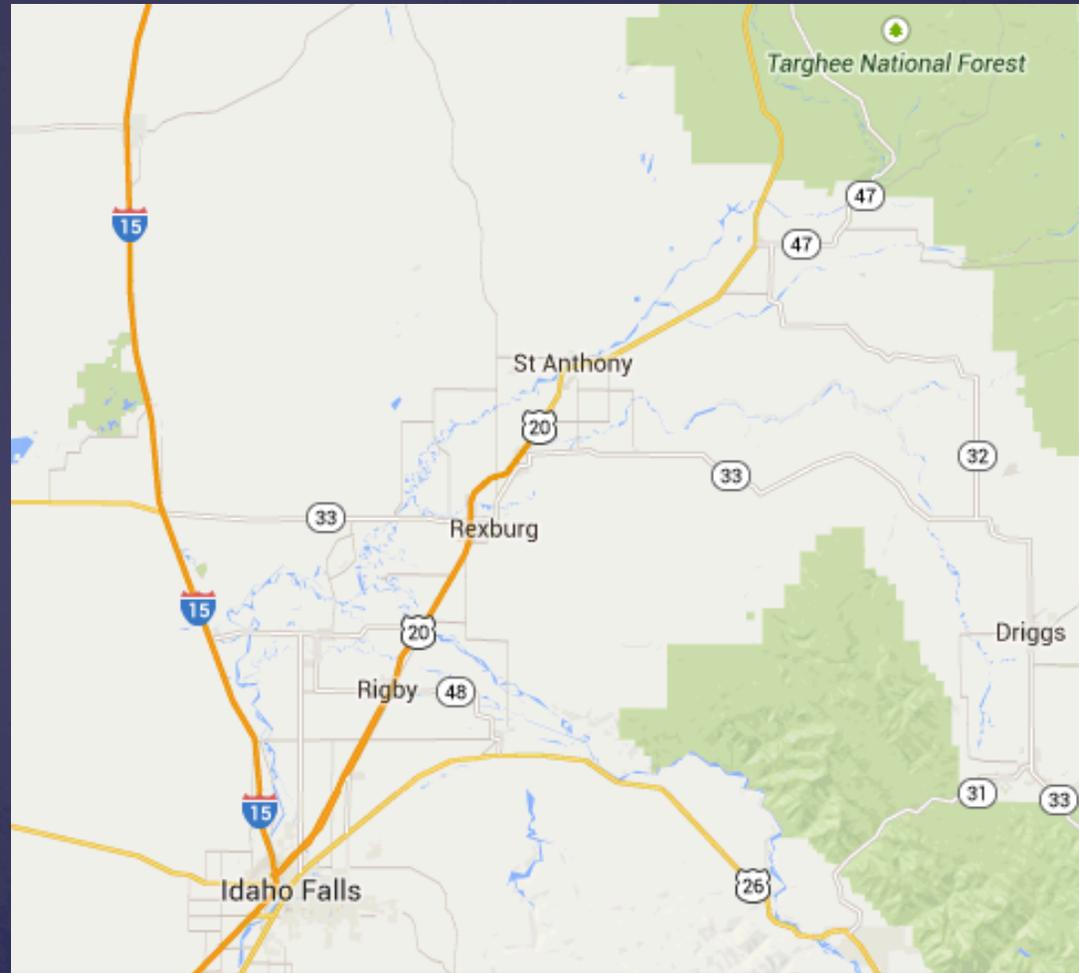


Source: Adapted from Arnold and Gibbons, 1996

Infiltration and Runoff

- ⌘ Different Factors such as soil types, slopes, land use, and imperviousness can greatly affect the quality of runoff in an urban area
- ⌘ The original paper focused on how the imperviousness nature of urban areas increases the amounts of Lead, Copper, and Zinc in Surface waters after a storm water runoff event

Water Quality Factors



Test Site: Snake River & Idaho Falls

Sample code number	99WA140	00WA155	00WA156	00WA157
Description	Tributary to Columbine Creek	Columbine Creek west headwaters	Columbine Creek east fork	Columbine Creek east fork
Date collected	9/22/1999	9/17/2000	9/17/2000	9/17/2000
Temperature (°C)	--- ¹	7.3	11.8	11.3
Density (g/mL) at 20°C	1.0024	0.99833	0.99835	0.99835
pH	2.70	6.68	6.32	3.89
Spec Cond (µS/cm) field / lab	--- / 820	58 / 59	136 / ---	221 / 241
Eh (V)	--- ¹	0.477	0.294	0.605
D.O. (mg/L)	--- ¹	8.11	7.8	7.8
<u>Constituent (mg/L)</u>				
Ca	6.1	4.4	8.9	10
Mg	2.9	2.3	6.8	7.5
Sr	0.090	0.051	0.079	0.097
Ba	0.026	0.021	0.019	0.025
Na	4.5	2.2	4.3	5.3
K	4.0	1.4	1.6	2.3
Li	0.002	<0.008	<0.008	<0.008
SO ₄	180	8.5	35	89
H ₂ S	--- ¹	<0.001	---	---
Alkalinity (as HCO ₃)	---	20.2	27.7	---

PHREEQC output tributaries to Snake River, WY

Pb		4.827e-09				
PbCO3		3.873e-09	3.874e-09	-8.412	-8.412	0.000
PbOH+		4.864e-10	4.712e-10	-9.313	-9.327	-0.014
Pb+2		3.531e-10	3.110e-10	-9.452	-9.507	-0.055
PbHCO3+		7.840e-11	7.595e-11	-10.106	-10.119	-0.014
Pb(OH)2		1.424e-11	1.425e-11	-10.846	-10.846	0.000
PbSO4		1.330e-11	1.330e-11	-10.876	-10.876	0.000
Pb(CO3)2-2		7.921e-12	6.975e-12	-11.101	-11.156	-0.055
PbCl+		2.628e-13	2.545e-13	-12.580	-12.594	-0.014
Pb(OH)3-		1.312e-14	1.271e-14	-13.882	-13.896	-0.014
Pb(SO4)2-2		6.033e-15	5.312e-15	-14.219	-14.275	-0.055
PbCl2		1.879e-17	1.880e-17	-16.726	-16.726	0.000
Pb2OH+3		4.367e-18	3.280e-18	-17.360	-17.484	-0.124
Pb(OH)4-2		2.570e-18	2.263e-18	-17.590	-17.645	-0.055
PbCl3-		4.499e-22	4.359e-22	-21.347	-21.361	-0.014
PbCl4-2		6.720e-27	5.917e-27	-26.173	-26.228	-0.055
PbNO3+		9.263e-36	8.973e-36	-35.033	-35.047	-0.014
Pb(2)		0.000e+00				

- ⌘ The waters typically feature low pH's
- ⌘ Feature an alkalinity around 20
- ⌘ Minerals containing Lead Saturation Index is Low
- ⌘ What happens when you introduce pollutants, specifically Lead, that are typical to the amounts found in storm water runoff in urban areas?

Initial Results

- ⌘ The Nationwide Urban Runoff Program
- ⌘ NURP discovered the average urban pollution amounts for Lead, Copper, and Zinc, as well as other water quality factors
- ⌘ These amounts found in major rivers and lakes at test sites around the nation close to urban environments

NURP

Pollutant	Units	Residential		Mixed		Commercial	
		Median	COV	Median	COV	Median	COV
BOD	mg/l	10	0.41	7.8	0.52	9.3	0.31
COD	mg/l	73	0.55	65	0.58	57	0.39
TSS	mg/l	101	0.96	67	1.14	69	0.85
Total Lead	$\mu\text{g/l}$	144	0.75	114	1.35	104	0.68
Total Copper	$\mu\text{g/l}$	33	0.99	27	1.32	29	0.81
Total Zinc	$\mu\text{g/l}$	135	0.84	154	0.78	226	1.07
Total Kjeldahl Nitrogen	$\mu\text{g/l}$	1900	0.73	1288	0.50	1179	0.43
Nitrate + Nitrite	$\mu\text{g/l}$	736	0.83	558	0.67	572	0.48
Total Phosphorus	$\mu\text{g/l}$	383	0.69	263	0.75	201	0.67
Soluble Phosphorus	$\mu\text{g/l}$	143	0.46	56	0.75	80	0.71

COV: Coefficient of Variation

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-----Description of solution-----
          pH = 8.007          Charge b
          pe = 7.775
    Activity of water = 1.000
      Ionic strength = 8.412e-04
    Mass of water (kg) = 1.000e+00
Total alkalinity (eq/kg) = 3.525e-04
    Total CO2 (mol/kg) = 3.311e-04
    Temperature (deg C) = 7.300
    Electrical balance (eq) = 2.462e-18
100*(Cat-|An|)/(Cat+|An|) = 0.00
          Iterations = 10
          Total H = 1.110128e+02
          Total O = 5.550765e+01

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Adding Urban Runoff Pollution Values to the Snake River

Pb	5.502e-07					
PbCO3	4.510e-07	4.511e-07	-6.346	-6.346	0.000	
PbOH+	5.635e-08	5.458e-08	-7.249	-7.263	-0.014	
Pb+2	3.130e-08	2.756e-08	-7.504	-7.560	-0.055	
PbHCO3+	6.985e-09	6.766e-09	-8.156	-8.170	-0.014	
Pb(OH)2	2.157e-09	2.157e-09	-8.666	-8.666	0.000	
Pb(CO3)2-2	1.213e-09	1.067e-09	-8.916	-8.972	-0.055	
PbSO4	1.178e-09	1.178e-09	-8.929	-8.929	0.000	
PbCl+	2.329e-11	2.255e-11	-10.633	-10.647	-0.014	
Pb(OH)3-	2.598e-12	2.516e-12	-11.585	-11.599	-0.014	
Pb(SO4)2-2	5.341e-13	4.702e-13	-12.272	-12.328	-0.055	
Pb2OH+3	4.486e-14	3.367e-14	-13.348	-13.473	-0.125	
PbCl2	1.665e-15	1.665e-15	-14.779	-14.778	0.000	
Pb(OH)4-2	6.652e-16	5.856e-16	-15.177	-15.232	-0.055	
PbCl3-	3.987e-20	3.862e-20	-19.399	-19.413	-0.014	
PbCl4-2	5.955e-25	5.242e-25	-24.225	-24.281	-0.055	
PbNO3+	4.095e-33	3.967e-33	-32.388	-32.402	-0.014	

The Saturation Indexes for Minerals containing Lead is a lot higher

- ‡ From the PHREEQ output, the pH jumped from 6.68 to over 8 and the saturation of lead increased, showing that storm water can change the geochemistry of natural rivers and streams in urban environments
- ‡ This can lead to Lead contaminated organisms
- ‡ However, This change is short term in nature, and can be influenced by a number of factors such as location of testing site at the urban environment, volume of the river, and amount of runoff

Results

⌘ [http://wwwbrr.cr.usgs.gov/projects/
GWC_chemtherm/pubs/ofr%2002-382.pdf](http://wwwbrr.cr.usgs.gov/projects/GWC_chemtherm/pubs/ofr%2002-382.pdf)

⌘ [http://water.epa.gov/scitech/wastetech/guide/
stormwater/upload/
2006_10_31_guide_stormwater_usw_b.pdf](http://water.epa.gov/scitech/wastetech/guide/stormwater/upload/2006_10_31_guide_stormwater_usw_b.pdf)

Sources