

A photograph of a vast, icy Antarctic landscape. In the foreground, a hiker with a large red backpack stands on a dark, rocky ridge, looking out over a wide, snow-covered valley. The background features rugged, snow-dusted mountains under a clear blue sky. The title text is overlaid in a large, red, serif font.

# The Whole Rock Chemistry and Mineralogy of Antarctic Cobbles

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NDSU Petrology Geol 422  
May 1, 2014

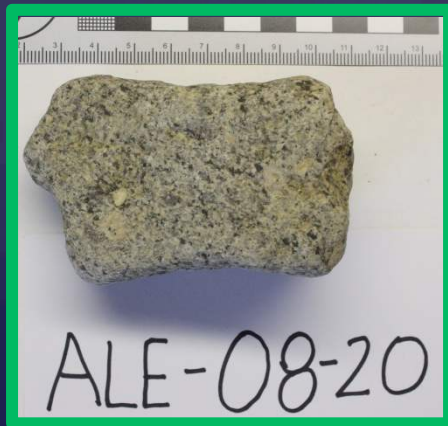
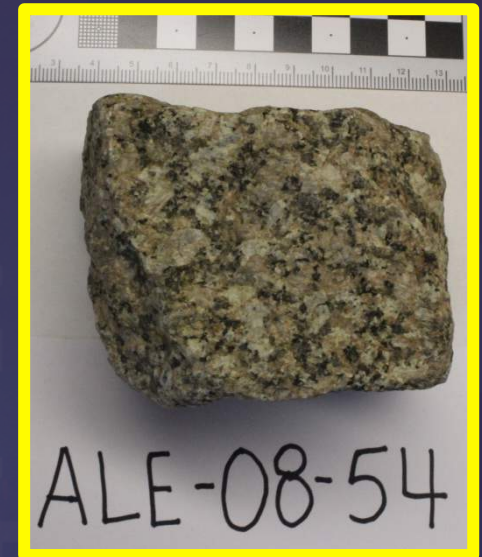
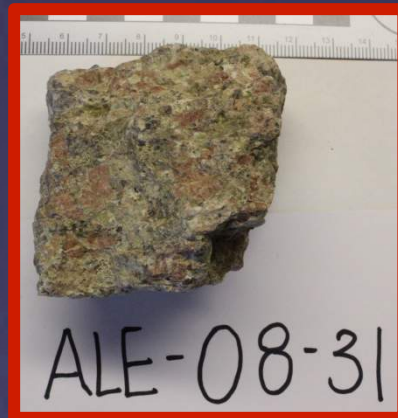
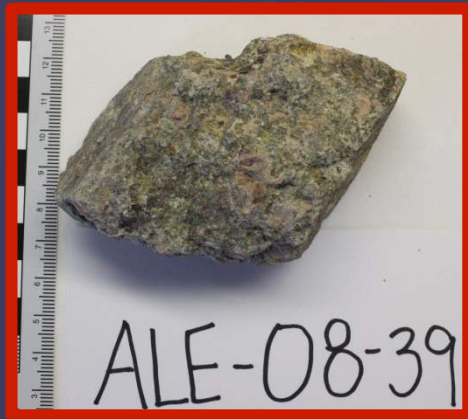
<http://people.bu.edu/marchant/Pictures/Dry%20Valleys.html>

# A Cobble's Journey

- Cobble's Back Story
- The Journey
  - From rock to powder
  - From powder to glass
- Unlocking K-Spar's Secret
- The Cobble's Conclusion



# Meet the Cobbles



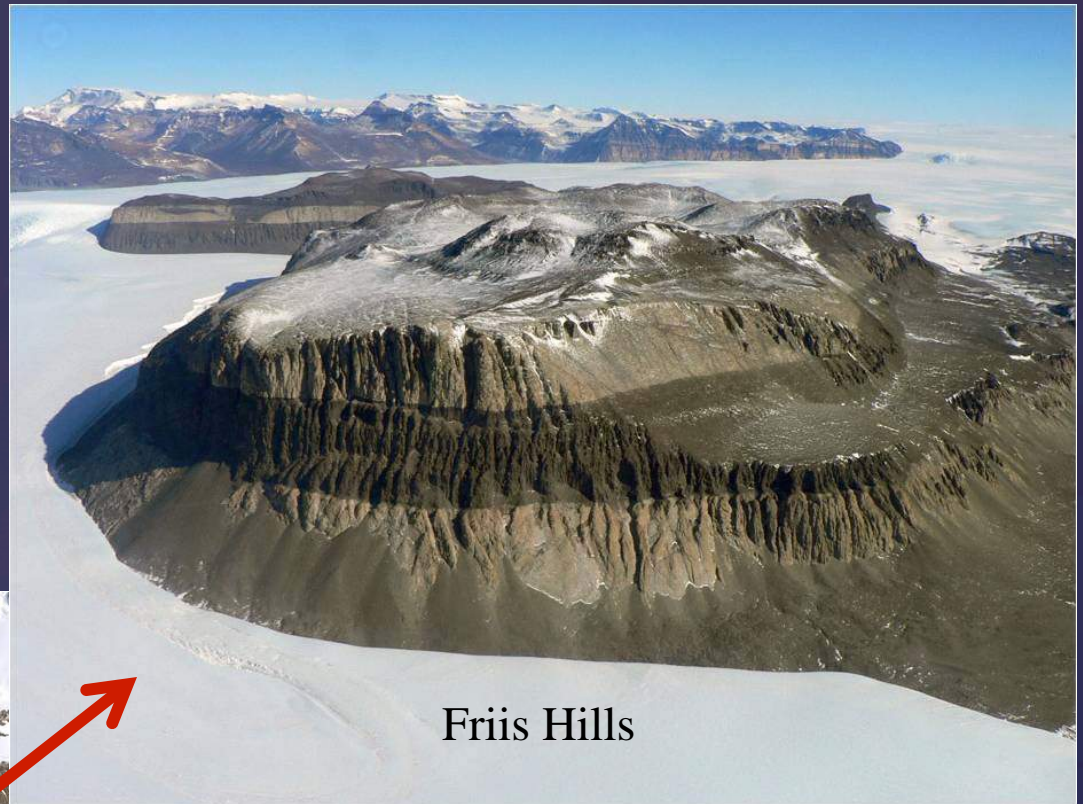
# Cobble's Back Story



Photos Courtesy of 2010 presentation by Kelsey Froward,  
Alex Smith, and Brian Hall



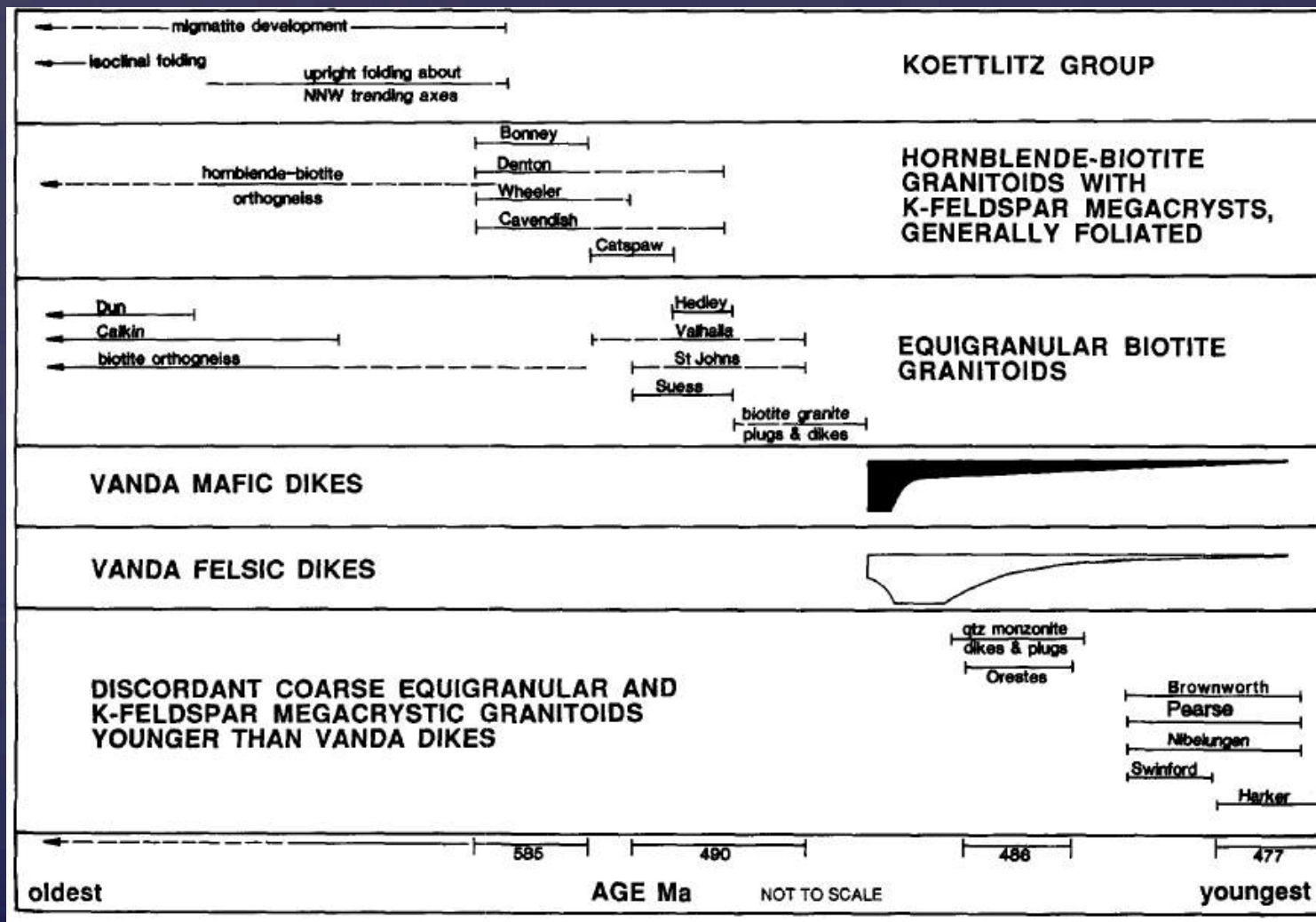
The Dry Valleys



Friis Hills

Photos Courtesy of 2010 presentation by Kelsey Froward, Alex Smith, and Brian Hall

# The Plutons of Antarctica



(Allibone et al, 1993)



# Possible Parents

- The Bonney, Cavendish, Denton, and Wheeler Plutons
  - Foliated, alkali megacrystals, hornblende, mafic enclaves
- Hedley, Valhalla, St. Johns, and Suess Plutons
  - Lack mafic enclaves, biotite, plagioclase, alkali
- Pearse Pluton
  - Alkali, plagioclase, hornblende, biotite
- Nibelungen Pluton
  - Alkali, porphyritic with aligned phenocrysts, chloritization

# Questions of the Cobbles

- What is the composition of the cobbles?
  - Using whole rock chemistry data from XRF
- What are the crystallization characteristics of the k-spar grains?
  - Using mineral data from XRD
- Can these cobbles be traced back to a specific pluton?

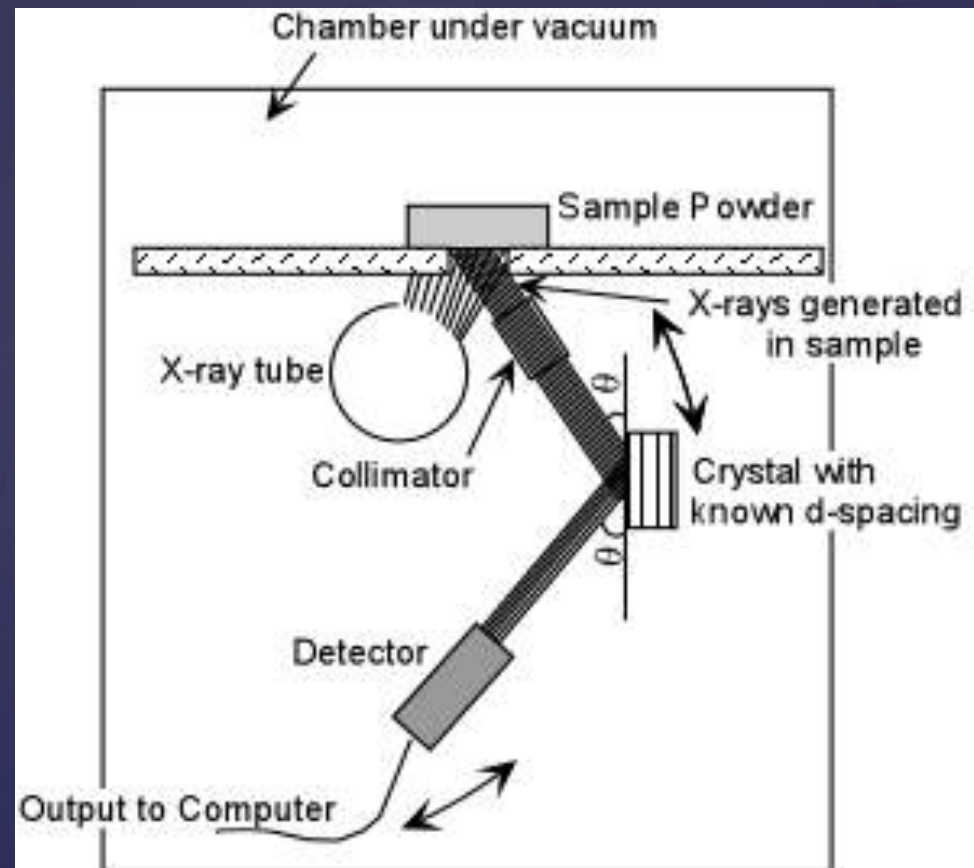


# What is XRF?

## X-Ray Fluorescence

Signal intensity generated by an element is compared to a standard of known composition to determine concentration

Signal is directly proportional to concentration in specimen



# From Rock to Powder

First the sample has to be cut and powdered

Sample can then be pressed into a powder pellet

This pellet is then placed in the XRF

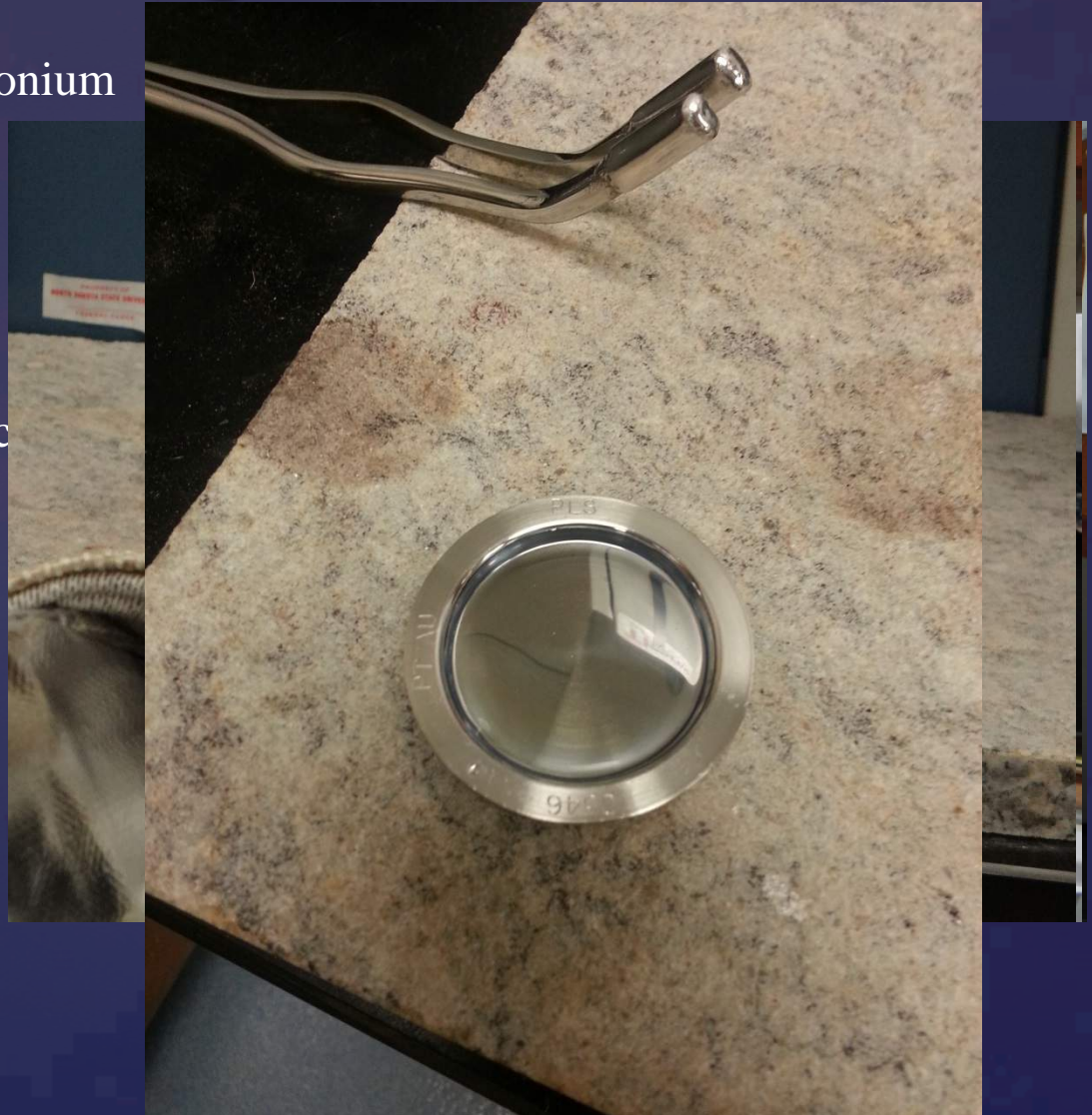
Photos Courtesy of Cheyanne Jacobs





# From powder to glass

- Mixed sample, flux, and Ammonium Nitrate
- Placed mixture in platinum crucible
- Melted sample in muffle furnace at 1100 °C for 5 minutes
- Swirled sample and added the mould to the furnace
- Heated for another 4 minutes
- Remove and pour immediately into mould



Photos Courtesy of Cheyanne Jacobs and Bernhardt Saini-Eidukat

# AFM Diagram

- Glass Discs
- Powder Discs

Calc-Alkaline

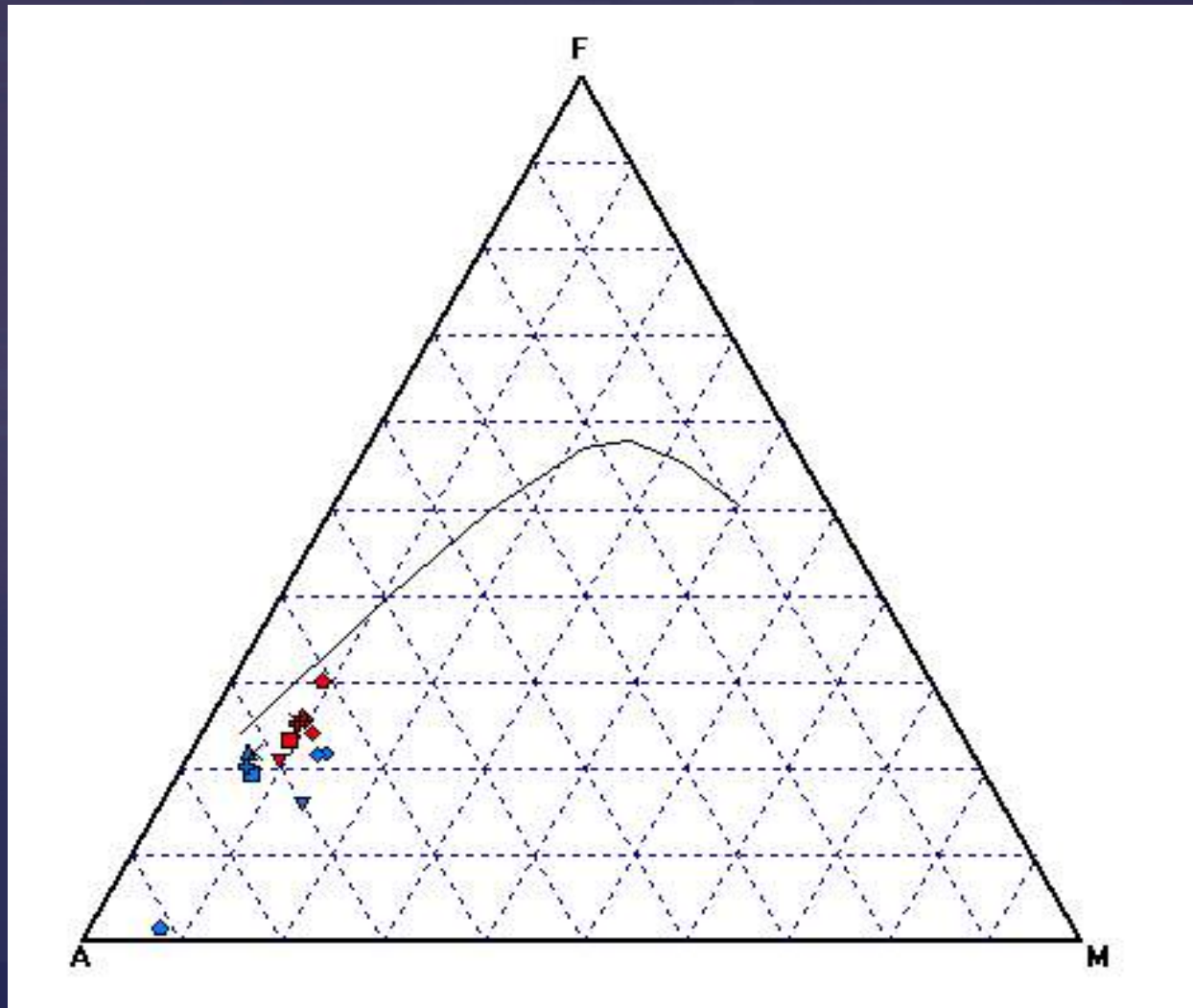


Diagram from Petrograph program



# TAS Diagram

## Samples

- ALE-08-39
- ALE-08-44
- ALE-08-36
- ALE-08-54
- ALE-08-46

Plotted in the  
syenite section

Glass samples  
tend to plot  
higher on the  
diagram than  
powder discs

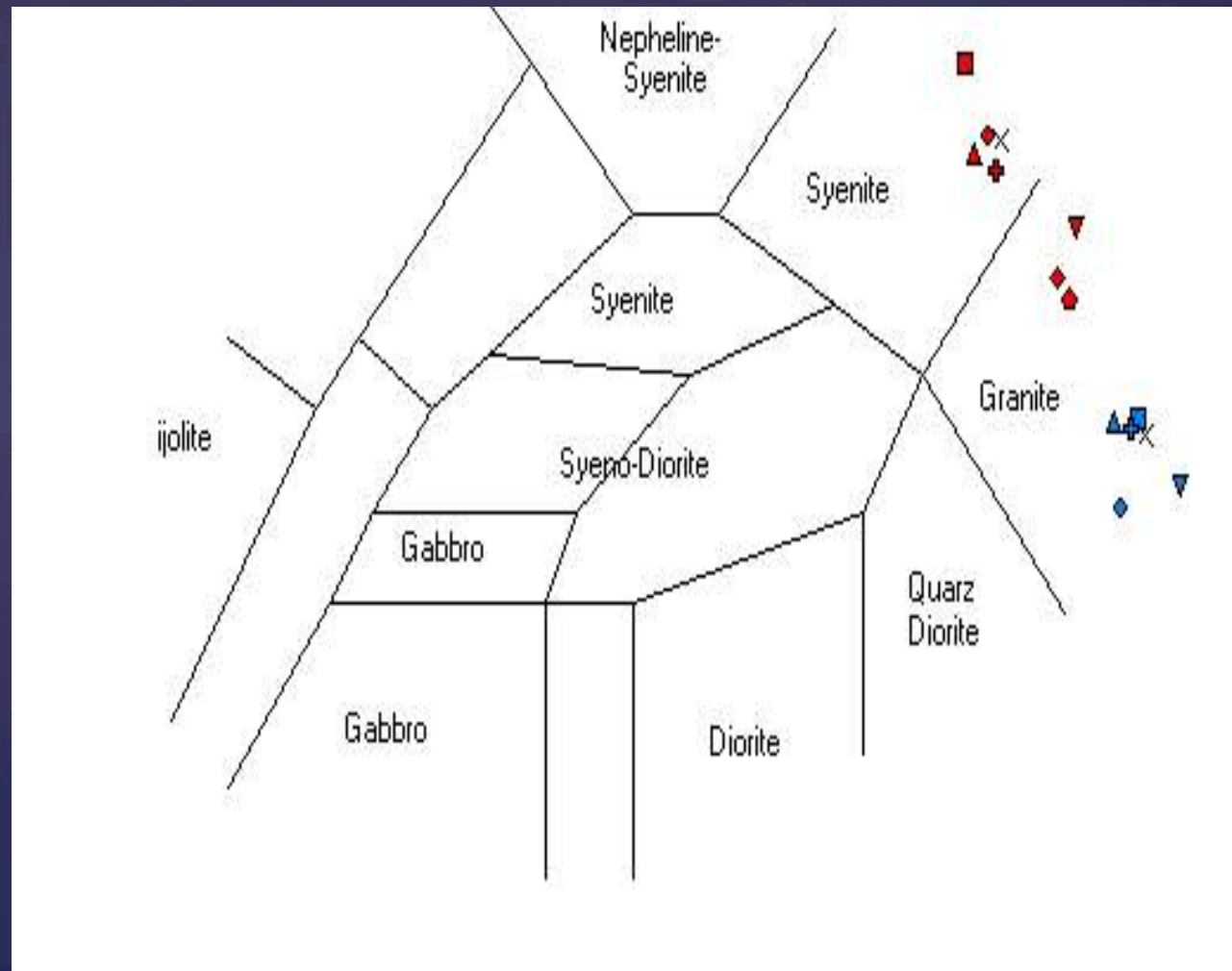


Diagram from Petrograph program

# Unlocking K-spars Secret





# K-spar Stats

Silicate Group	Tectosilicates (3-D framework)
Chemical Formula	$\text{KAlSi}_3\text{O}_8$
Varieties	Microcline Orthoclase Sanidine

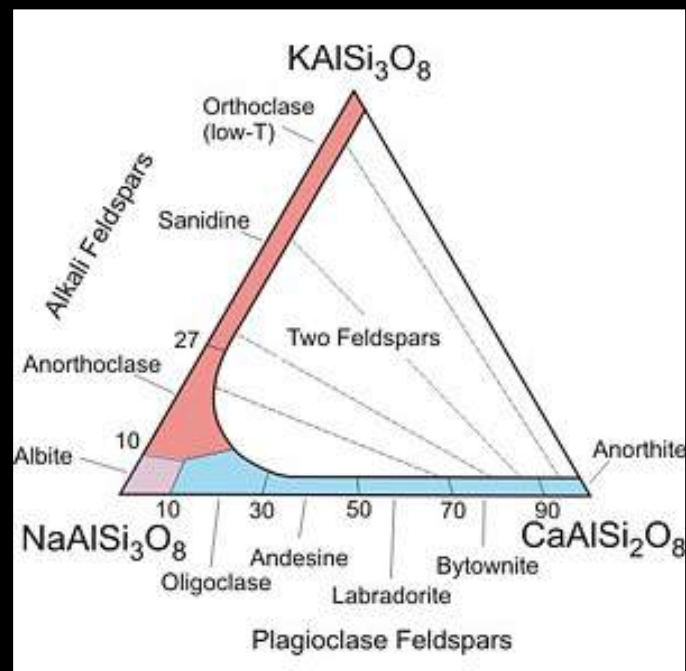


Diagram from: [www2.imperial.ac.uk](http://www2.imperial.ac.uk)

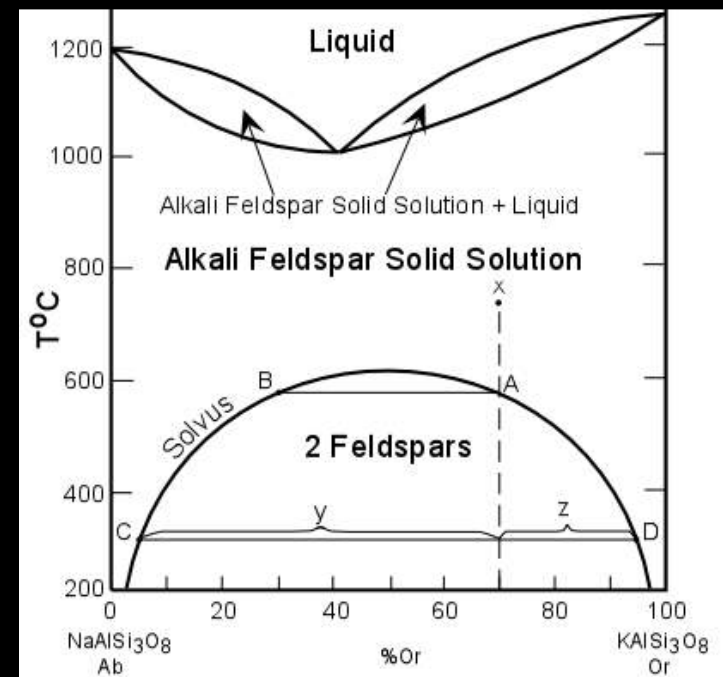


Diagram from: [www.tulane.edu/](http://www.tulane.edu/)

# What's K-spars secret?

- $\text{Al}^{3+}$  substitutes for  $\text{Si}^{4+}$  because of its similar size and charge
- The extent of substitution can be calculated using unit cell parameters measured from XRD
- Can give us information on crystallization setting, such as:
  - Amount of order/disorder
  - When the crystals formed during cooling
- Using this information alongside stratigraphy of rock units, depth of formation can be estimated



# Sample preparation

- Select grains from sample ALE-08-26
- Powder them in agate mortar & pestle
- Place in slide cavity
- Insert slide into x-ray powder diffraction machine

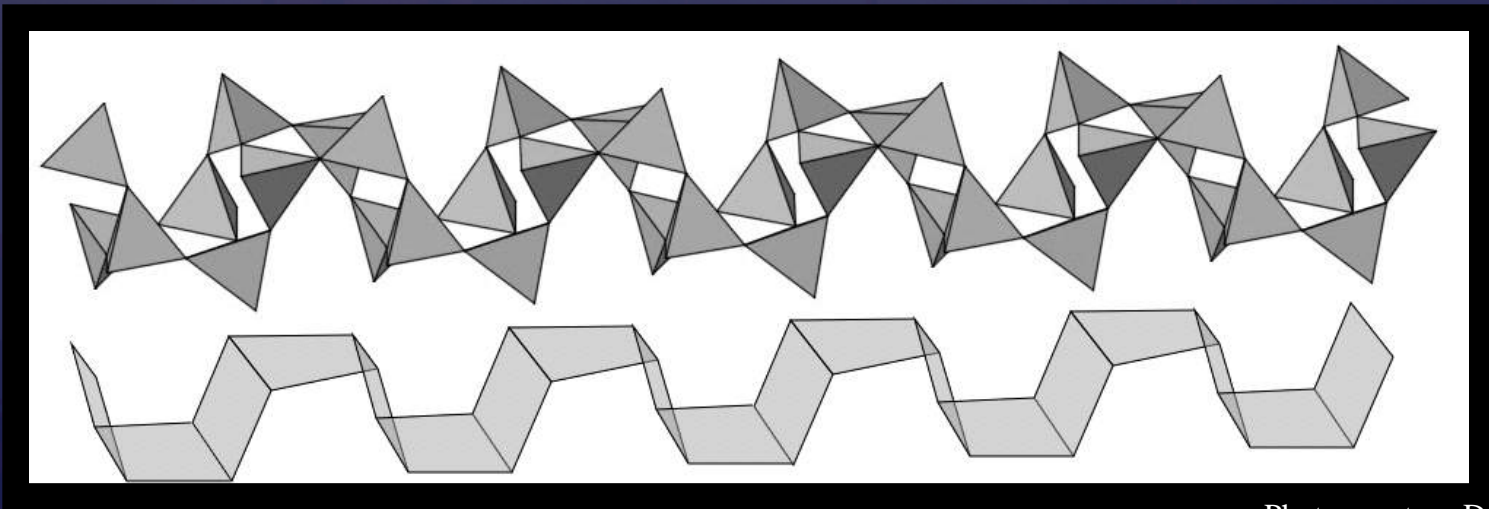


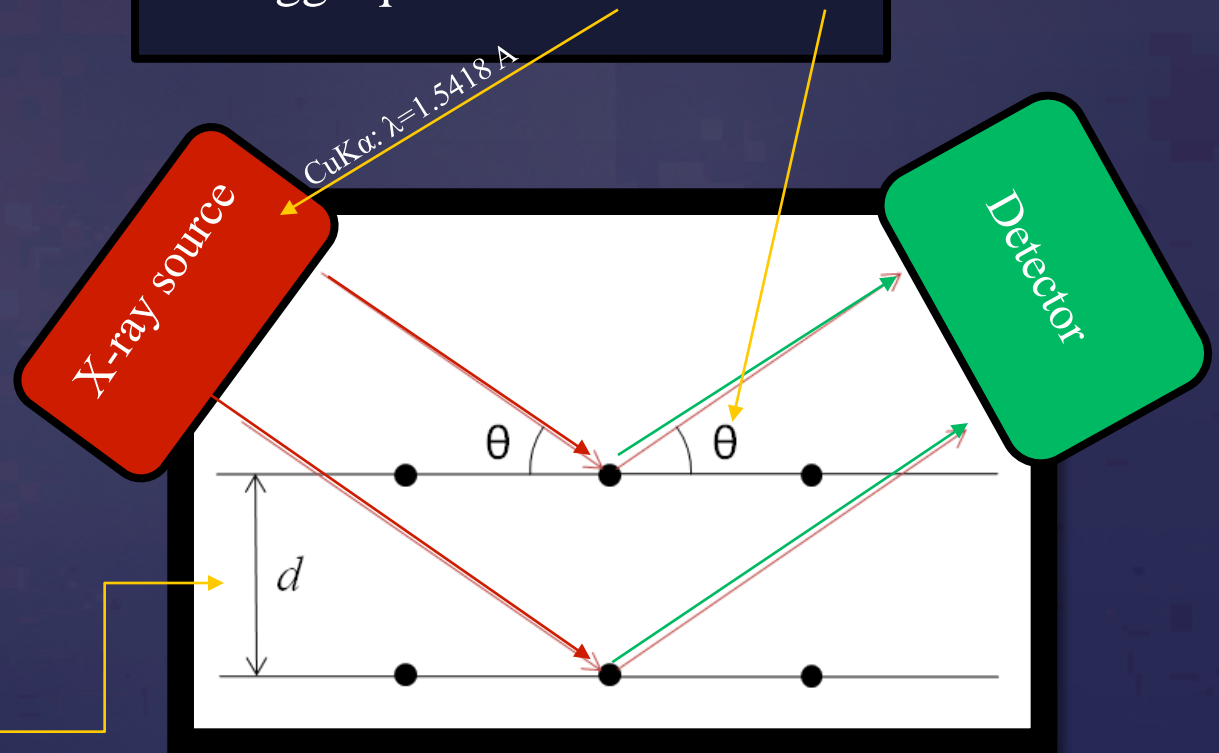
Photo courtesy Dr. Eidukat

# X-Ray Powder Diffraction

$$\text{Bragg equation: } n\lambda = 2d\sin\theta$$



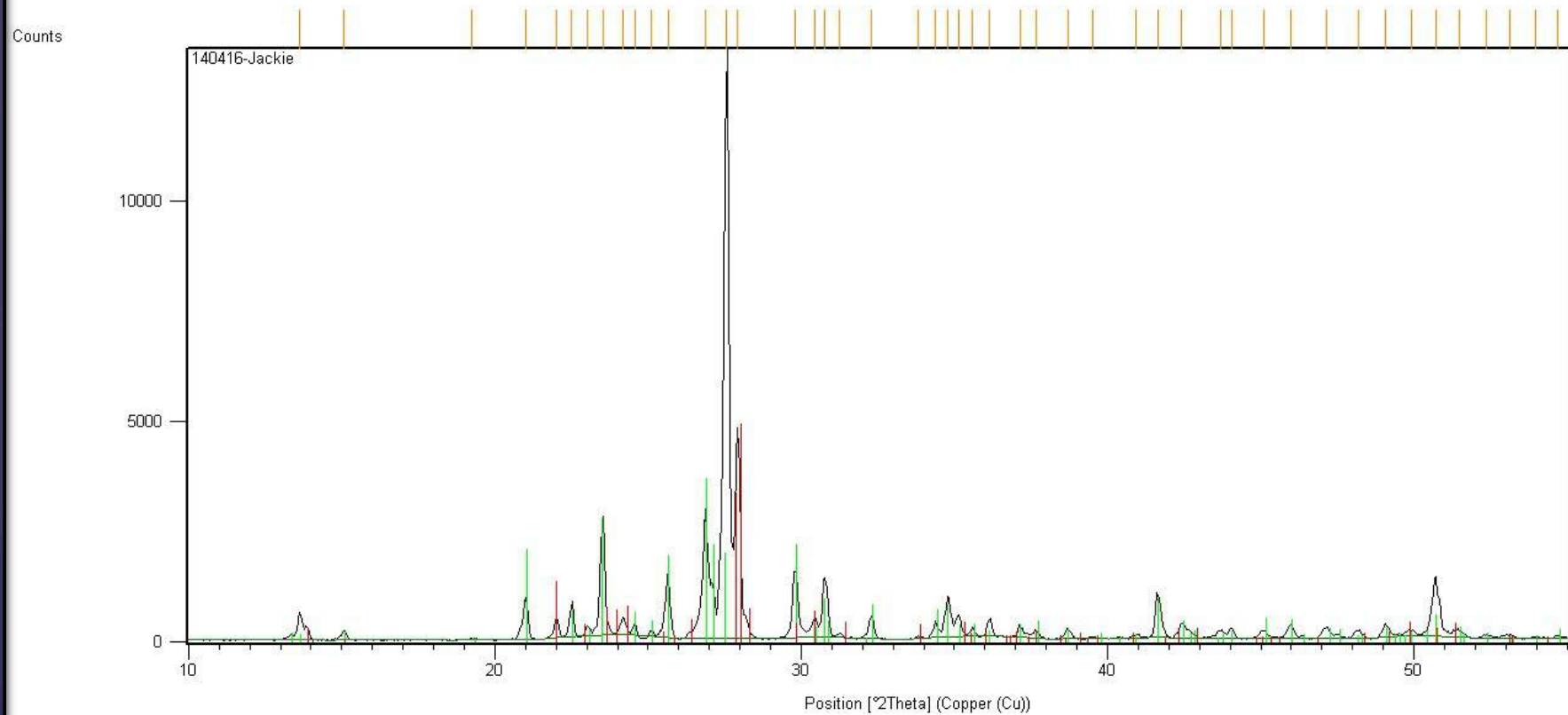
<http://serc.carleton.edu/>



<http://chemwiki.ucdavis.edu/>

D-spacing is unique to each mineral & is used for mineral identification & to determine cell parameters

# XRD Results



— Orthoclase  
— Albite

Photo courtesy Dr. Angel Ugrinov



# Comparing peaks

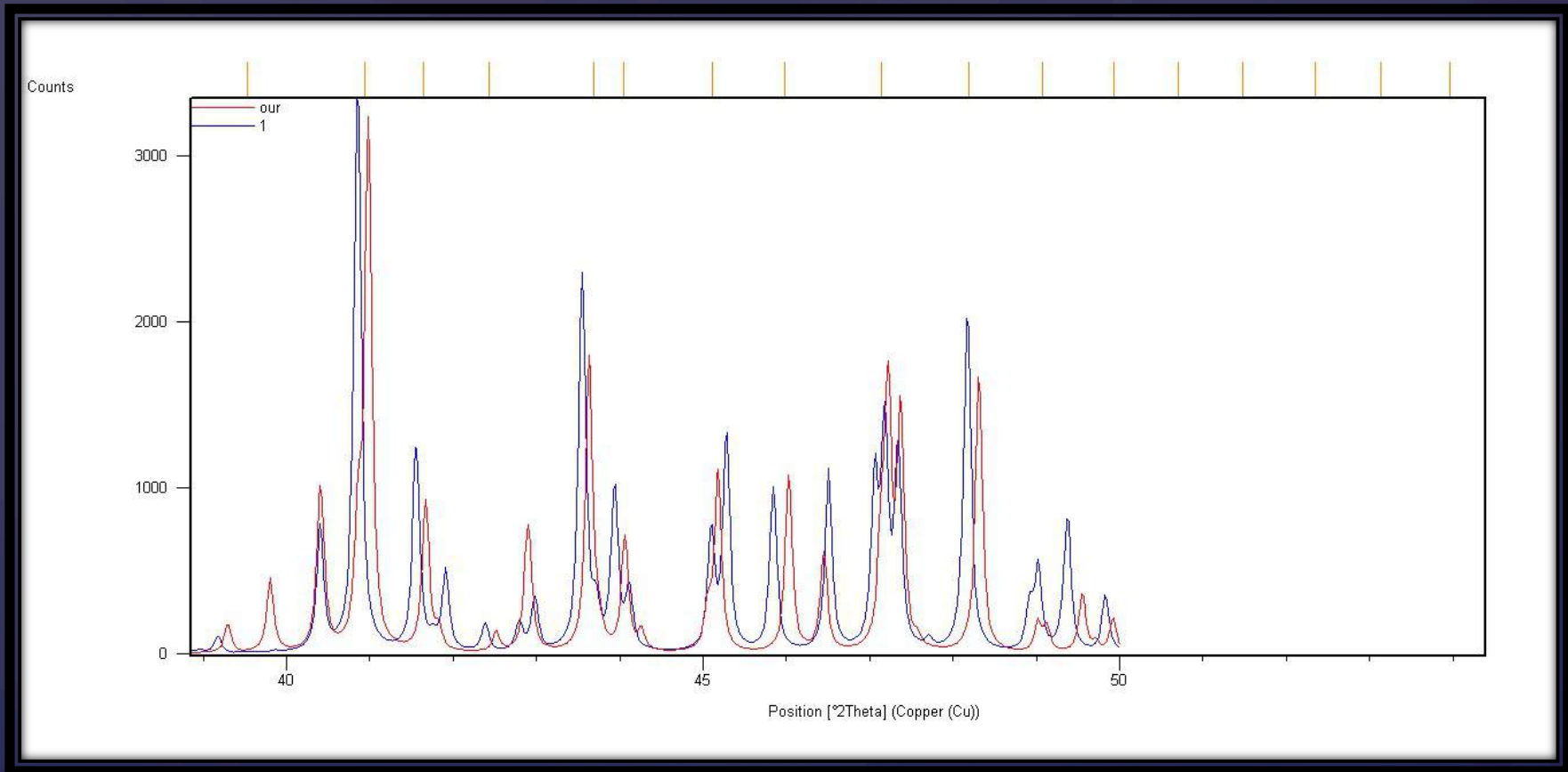
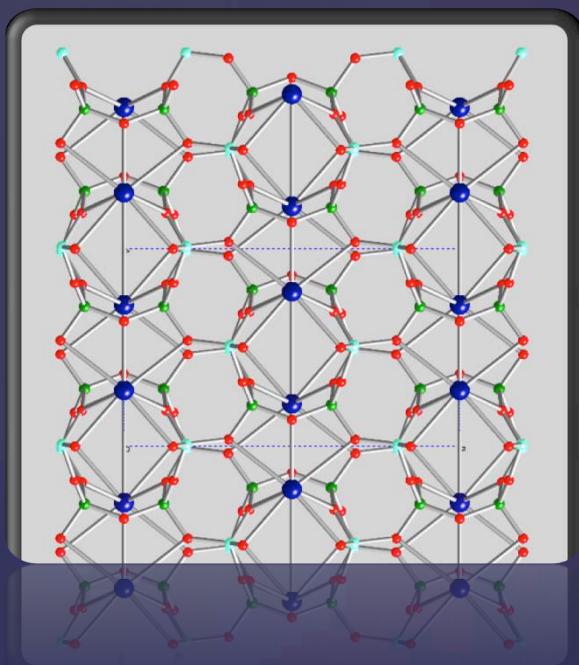


Photo courtesy Dr. Angel Ugrinov

- Orthoclase template that matched our peaks
- Other orthoclase template

# XRD Results

Photo courtesy Dr. Ugrinov



## Unit Cell Details

a	8.5610 Å
b	12.9960 Å
c	7.1920 Å
$\alpha$	90.0000°
$\beta$	116.0000°
$\gamma$	90.0000°
Crystal system	Monoclinic
Space group	C2/m
Unit cell volume ( $V=abc \cdot \sin\beta$ )	719.19 Å <sup>3</sup>

# Unlocking K-spar: Step 1

Determine  $^n\text{Or}$  – the % orthoclase compared to albite in crystal

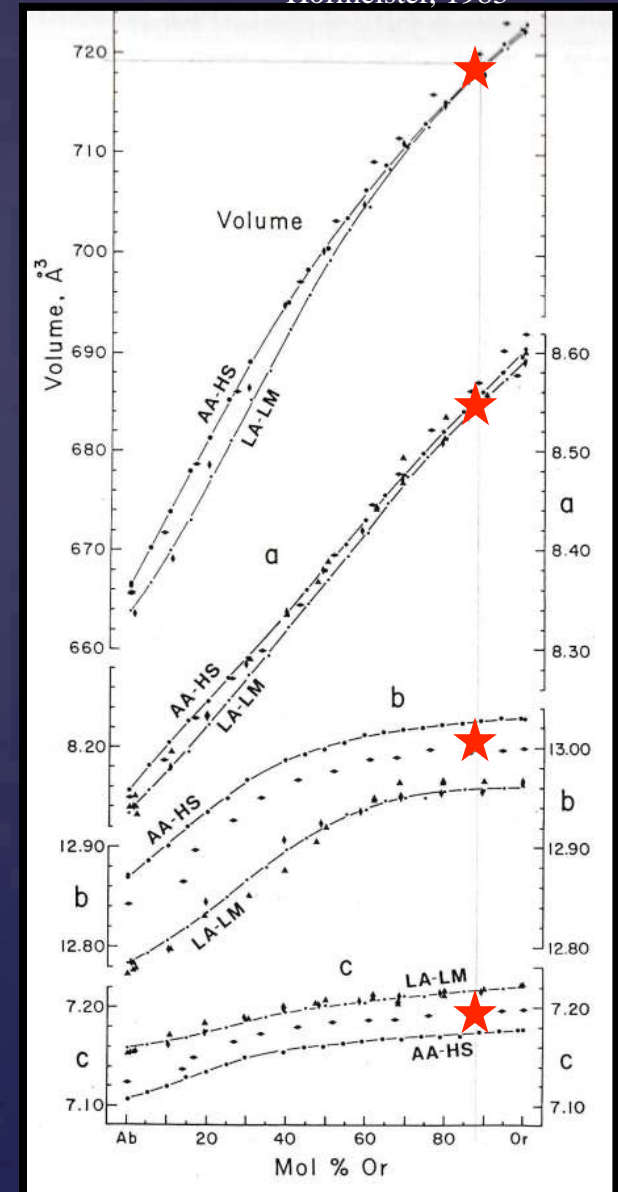
- Use equation from Ribbe & Hofmeister:

$$^n\text{Or} = (-584.6683) + (2.58732 V) - (3.83499 \times 10^{-3} V^2) + (1.90428 \times 10^{-6} V^3)$$

$$^n\text{Or} = 89.07$$

- Our cell parameters match AA-HS system the closest (Analbite-High Sanidine) when plotted on the graph

Photo from Ribbe & Hofmeister, 1983





# Using the b•c method to determine crystallization characteristics

b and c cell parameters depend on both composition & Al, Si order in alkali feldspars and plagioclases

From graph & equations  
from Ribbe &  
Hofmeister,

$$2t_1 = 0.7194$$

LA-LM series: fully  
ordered ( $2t_1 = 1.00$ )  
AA-HS series: fully  
disordered ( $2t_1 = 0.50$ )

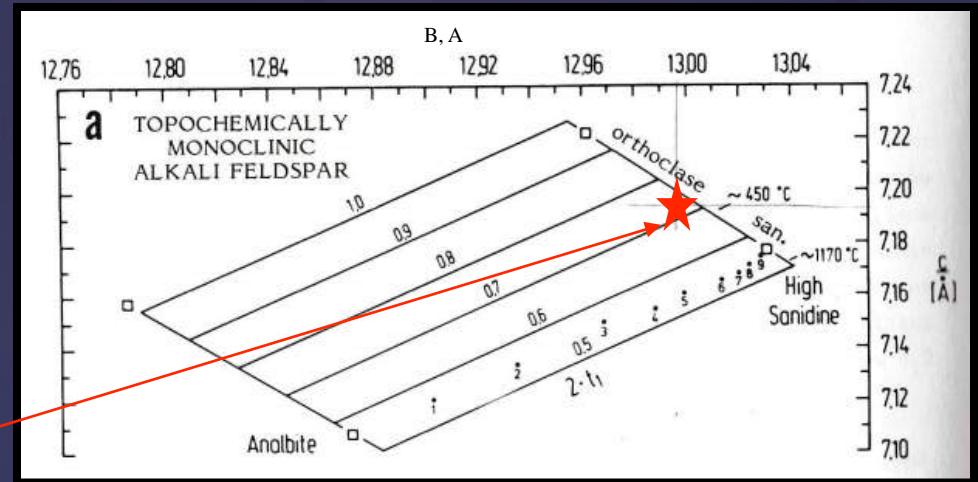


Photo from Ribbe & Hofmeister, 1983

From this, our sample is found to  
be slightly disordered.

# Conclusion from b•c method

Database predictions for T1 & T2 Al concentrations:

T1= 35% Al, 65% Si

T2= 15% Al, 85% Si

Al from T2 concentrated/migrated to T1 site, but no Al  
from  $t_{1m}$  migrated to  $t_{1o}$  site-

Therefore, our K-spar is relatively disordered &  
formed in the 1<sup>st</sup> half of crystallization.

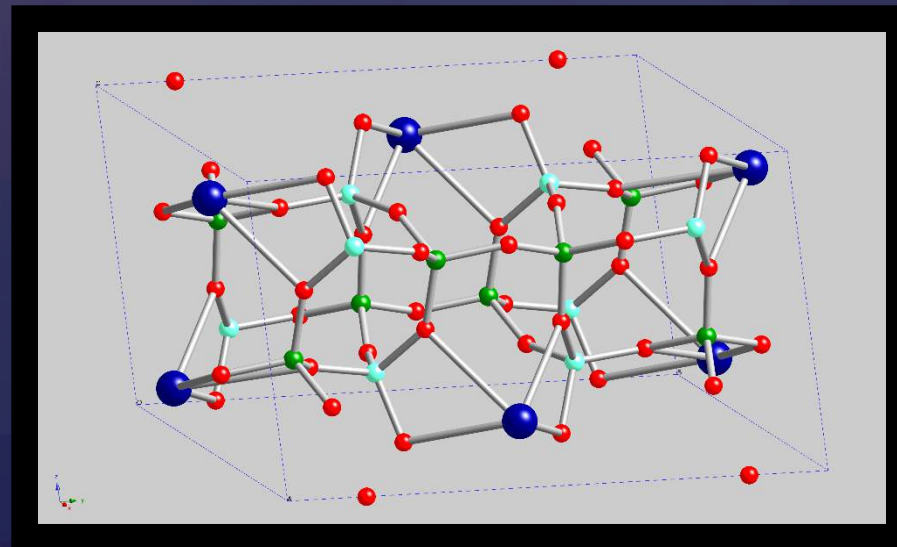


Photo courtesy Dr. Ugrinov

# The $[110]$ Method

- During ordering Al concentrates in the  $t_{1o}$  tetrahedra sites.
- 4-fold rings arranged along  $[110]$  become depleted
- In monoclinic feldspars the most contraction is along the  $b$  axis
- Translation distances along  $[110]$  and  $[1\bar{1}0]$  are measured to determine ordering

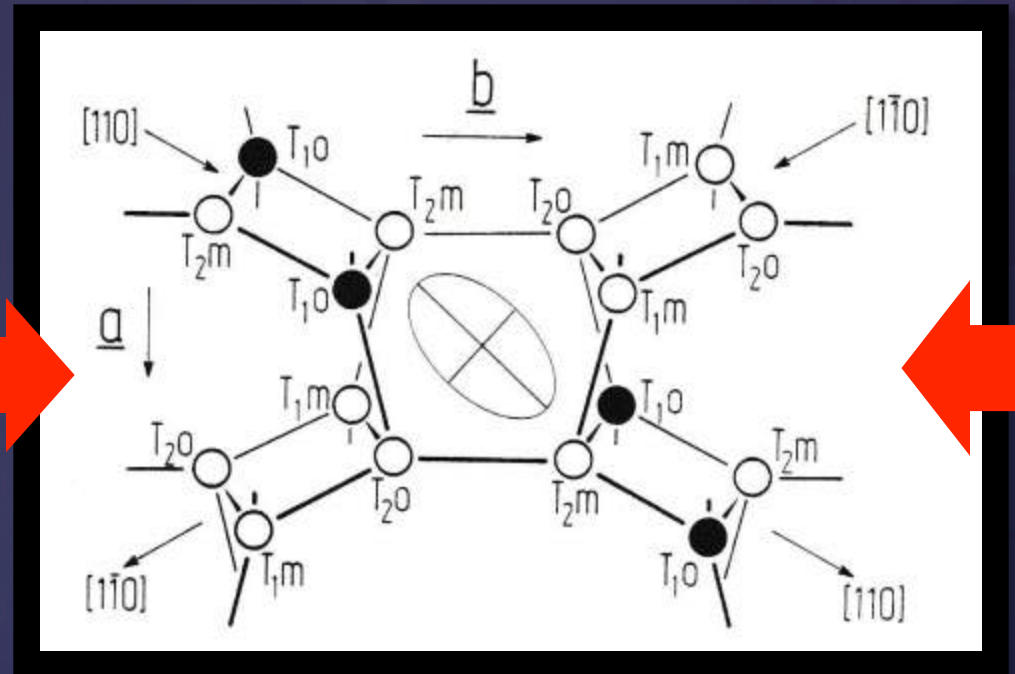


Photo from Ribbe & Hofmeister, 1983

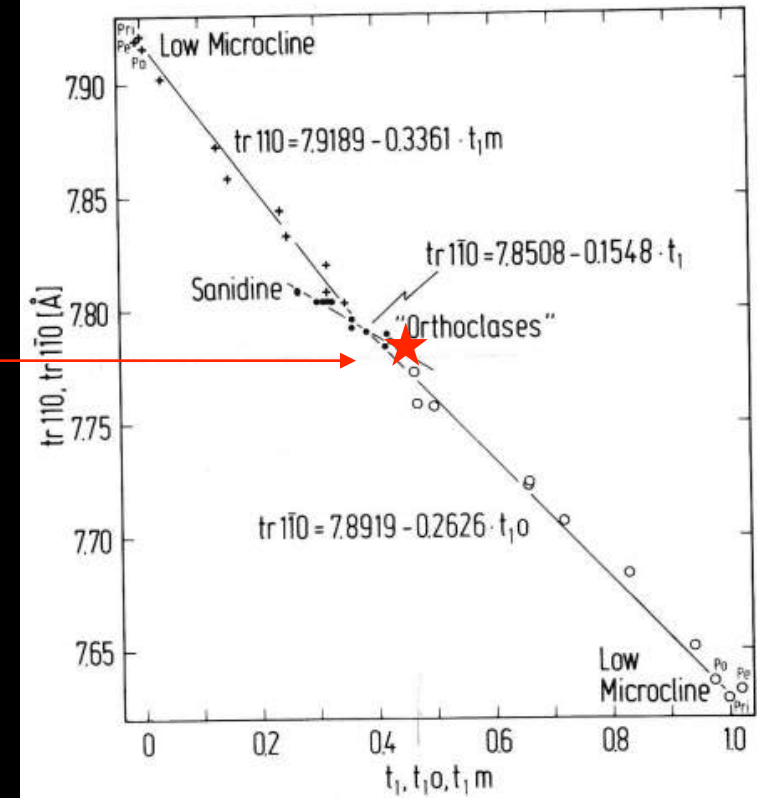
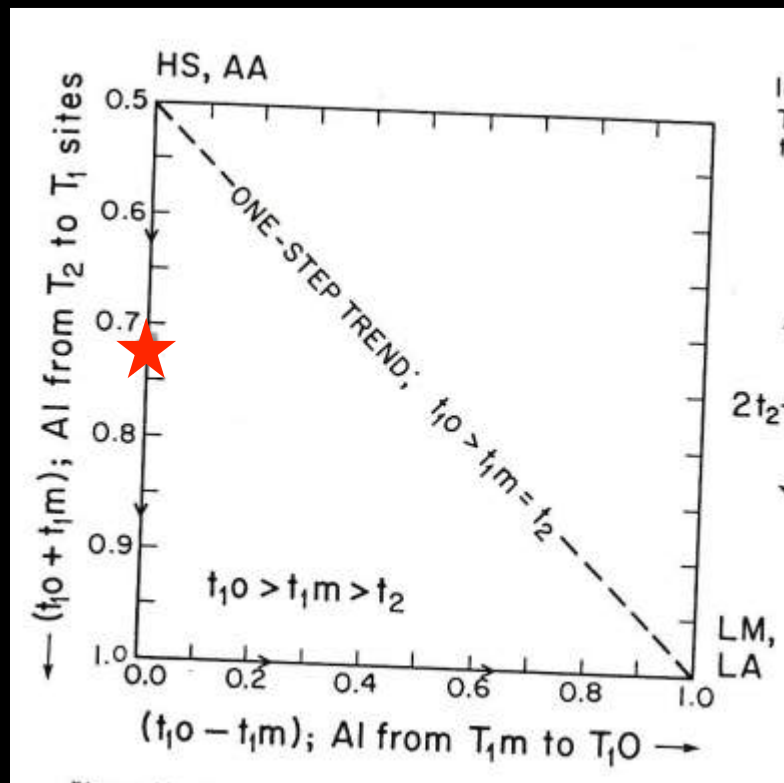


From Ribbe & Hofmeister equations:

$$\text{tr}[110] = 7.7812 \text{ \AA}$$

$$\text{tr}[110] = 7.7812 \text{ \AA}$$

$$T_1, t_1o, t_1m = 0.465$$



[110] Method conclusions:  
Relatively disordered crystal that formed in  
the 1<sup>st</sup> half of crystallization

# Cobble's Conclusion

What is the whole rock chemistry?

The samples are calc-alkaline

The powder pellet results show that the samples are all granitic

The glass bead results show most of the cobbles to be syenites

Crystallization Characteristics?

K-spar is relatively disordered and formed during first half of crystallization

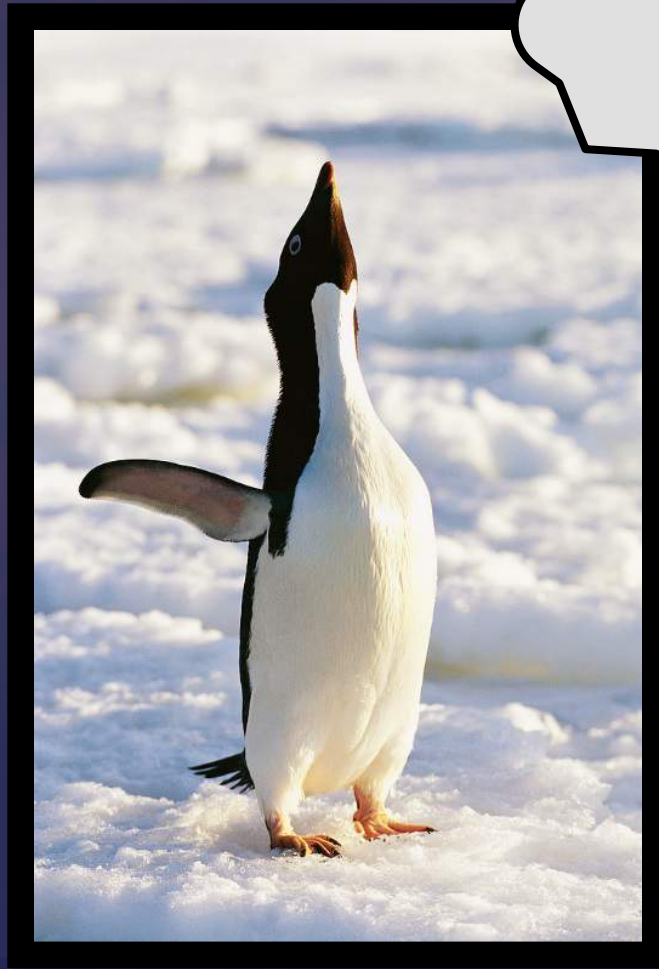
Is it possible to trace samples to a pluton?

**No**

With the data we collected, it is not possible to determine a parent pluton due to the striking similarities between plutons.

Although their secret was unlocked, the cobble's did not find their home...

# Thank you!



The cobble's shall  
never return home!!



# References

Allibone, A. H., Cox, S. C., Graham, I. J., Smillie, R. W., Johnstone, R. D., Ellery, S. G., Palmer, K., 1993. Granitoids of the Dry Valleys area, southern Victoria Land, Antarctica: plutons, field relationships, and isotopic dating. *New Zealand Journal of Geology and Geophysics* 36, 281-297.

Smillie, R. W., 1992. Suite subdivision and petrological evolution of granitoids from the Taylor Valley and Ferrar Glacier region, south Victoria Land. *Antarctic Science* 4, 71-87.

Ribbe, P. and Hofmeister, A., 1983. Feldspar Mineralogy. *Nature* 2,

Thank you

Dr. Eidukat

Dr. Lewis

Dr. Ugrinov

WERG Lab